TED	(10)	-1	01	6	B
31.75					

(REVISION—2010)

Reg. No.	
Signature	

SECOND SEMESTER DIPLOMA EXAMINATION IN ENGINEERING/ TECHNOLOGY—OCTOBER, 2012

APPLIED SCIENCE—II

Chemistry

(Common Except DCP & CABM)

[Time: 1½ hours

(Maximum marks: 50)

		Colonia High softe melt uten deuts teurs se par laim paler me uten la servicio de la colonia de la c	Marks
		PART—A	
	(A	nswer all questions in one or two sentences. Each question carries 2 marks.)	
Ι	(a)	Define the term Vulcanisation.	2
	(b)	What is Anodising?	- 2
		PART—B	
		(Answer any two questions. Each question carries 8 marks.)	
II	(a)	What are the factors that affect adsorption?	4
	(b)	Write the chemical reaction takesplace in H_2 - O_2 fuel cell and give the advantages of H_2 - O_2 fuel cells.	4
III	(a)	Explain the theory of electro chemical corrosion.	4
	(b)	What are the differences between thermoplastic and thermosetting plastic?	4
(V	(a)	Explain about the uniqueness of carbon.	4
	(b)	Suggest remedial measures in order to reduce green house effect.	4
		PART—C	
4		Answer one full question from each unit. Each question carries 15 marks.)	
8	20	Unit – I	
V	(a)	How will you distinguish between physical adsorption and chemical adsorption ?	4
	.(b)	Explain the mechanism of electrolysis by taking molten NaCl as an example.	4
	(c)	Explain the following:	
		(i) Galvanic corrosion (ii) Differential aeration corrosion.	4
	(d)	What are primary and secondary cell? Discuss one example of each.	3
		OR	

[77]

VI	(a)	List the important application of adsorption.	lark:
		(i) Why does blue colour of copper sulphate solution get discharged when iron rod is dipped into it?	
		(ii) Calculate the EMF of the cell given:	
		$E^{\circ} Cu^{2+}/Cu = +0.34V$; $E^{\circ} Fe^{2+}/Fe = -0.44 V$.	4
	(c)	Explain cathodic protection and Barrier protection method inorder to prevent corrosion.	4
	(d)	What are the differences between metallic conduction and electrolytic conduction?	3
		Unit – II	
VII	(a)	Define the following terms: (i) Functional group (ii) Isomerism.	3
	(b)	What are Homopolymer and copolymer and give two example each?	4
		What are the qualities of a good fuel?	4
	(d)	Explain the following: (i) Green house effect (ii) Acid rain.	4
		OR	
VIII	(a)	Differentiate between saturated and unsaturated compounds.	4
	(b)	Define the term composite. Explain the different types of composites.	4
	(c)	Write the monomers of the following polymers:	
		(i) Buna-s (ii) Neoprene (iii) Teflon (iv) Nylon 6.	4
	(d)	What is Green Chemistry? Explain the scope of Green chemistry in the present world.	3
	1		
	10		
	F 12	parties because the subjection between process despited, and the ways but the	